Domino Reactions

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Palladium-Catalyzed Three-Component Synthesis of 3-(Diarylmethylene)oxindoles through a Domino Sonagashira/ Carbopalladation/C-H Activation/C-C Bond-Forming Sequence**

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Devising novel multicomponent reactions (MCRs) that achieve the formation of multiple bonds in one operation is one of the major challenges in modern organic synthesis.^[1] As such processes avoid time-consuming and costly purification processes, as well as protection-deprotection steps, they are inherently environmentally benign and atom economic.[2] Whereas some powerful MCRs such as the Passerini^[3] and Ugi^[4] reactions proceed in the absence of external reagents. most chemical transformations involve reactants that are not active enough to be self-assembled. Thus, the use of catalytic rather than stoichiometric amounts of external reagents to trigger the reaction is highly desirable to minimize the production of waste.^[5] As a continuation of our ongoing project on the development of palladium-catalyzed domino processes^[6] as well as direct C-H functionalization,^[7] we report herein a palladium-catalyzed three-component synthesis of 3-(diarylmethylene)indolinones.[8-13] The underlying principle of our approach is shown in Scheme 1. The Sonogashira coupling^[14] of *N*-aryl-*N*-alkyl propiolamides **1** with an aryl iodide 2 should give the phenyl propiolamide 5, which would then react with a second aryl halide 3, ideally in the presence of the same catalyst, to afford the target compound 4 through a sequence of intermolecular carbopalladation, C-H activation, and a C-C bond-forming process. $^{[15,16]}$ Whereas many palladium-catalyzed transformations have been developed and widely used in novel domino processes for the syntheses of heterocycles. [17,18] the MCR that combines mechanistically distinct reactions by a single catalyst is by far less developed owing to the specificity of each catalytic system to each individual reaction.^[19] To our knowledge, the use of a single palladium catalyst to catalyze three different reactions, namely aryl alkynylation, carbopal-

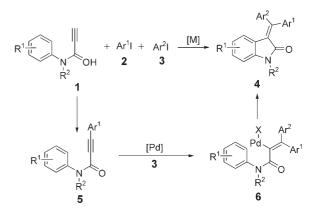
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Scheme 1. Proposed three-component reaction to synthesize 3-(diarylmethylene) oxindoles.

ladation, and direct C-H functionalization, has not been reported previously.[20]

Very few examples have been reported that deal with the successful Sonogashira reaction of electron-deficient alkynes, such as propiolic esters^[21] and propiolamide,^[22] with aryl halides. Despite this potential pitfall, we performed a survey of reaction conditions using N-(4-methoxyphenyl)-N-methyl propiolamide (1a), phenyl iodide (2a), and 2-nitrophenyliodide (3a) as test substrates. In the event, stirring a solution of 1a and 2a in N,N-dimethylformamide (DMF) in the presence of [Pd(PPh₃)₄] (5 mol %), copper iodide (15 mol %), and Et₃N at 60°C for 1 h followed by addition of 3a and heating to 110°C for 15 h afforded the expected oxindole 4a in 28% yield. Control experiments indicated that a) Sonogashira coupling between 1a and 2a, the presumed first step of this domino process, proceeded smoothly to furnish the phenylpropiolamide 5a in over 90% yield, b) copper iodide is required for the successful Sonogashira coupling between 1a and 2a, [23] and c) the Sonogashira coupling between 1a and 2a proceeded sluggishly under ligand-free conditions, although such conditions were of choice for the carbopalladation/C-H functionalization step. With these results in hand, we set out to focus on the overall reaction efficiency of this novel threecomponent reaction instead of optimizing the individual steps^[24] by varying the palladium source, the ligand, the amount of copper, the base, and the solvent. The results are summarized in Table 1.

The base played an important role in the overall efficiency of this three-component reaction. Simply by switching from triethylamine to sodium acetate under otherwise identical conditions furnished the desired oxindole 4a in 73% yield.

Table 1: Palladium-catalyzed three-component synthesis of oxindoles: a survey of reaction conditions. [a]

Entry	[Pd] (mol%)	Cul [mol%]	Base (equiv)	Solvent	t_1, t_2 [h]	Yield [%]		
•	,		,			4 a	5a Š	4 b
1	[Pd(PPh ₃) ₄] (5)	15	Et ₃ N (2)	DMF	1, 15	28	-	
2	$[Pd(PPh_3)_4]$ (5)	15	NaOAc (2)	DMF	4, 45	73	11	3
3	Pd(OAc) ₂ /PPh ₃ (5:10)	15	NaOAc (2)	DMF	4, 17	58	_	6
4	$[Pd(PPh_3)_2Cl_2]$ (5)	15	NaOAc (2)	DMF	4, 3	48	_	3
5	[Pd(dba)]/PPh ₃ (5:10)	15	NaOAc (2)	DMF	1, 13	58	_	10
6	[Pd(PPh ₃) ₄] (5)	15	NaOAc (2)	DMA	1, 42	53	5	8
7	$[Pd(PPh_3)_4]$ (5)	15	NaOAc (2)	DMSO	1, 42	52	46	_
8	$[Pd(PPh_3)_4]$ (5)	15	NaOAc (2)	Dioxane	17, 48	-	n.d. ^[b]	_
9 ^[c]	$[Pd(PPh_3)_4]$ (10)	5	NaOAc (2)	DMF	1, 39	-	6	79
10 ^[c]	$[Pd(PPh_3)_4]$ (5)	5	KOAc (2)	DMF	1, 24	79	13	_
11 ^[c]	$[Pd(PPh_3)_4]$ (10)	5	K_2CO_3 (2)	DMF	1, 28	66	7	11
12 ^[c]	$[Pd(PPh_3)_4]$ (10)	5	NaHCO ₃ (2)	DMF	1, 39	63	6	3
13 ^[c]	$[Pd(PPh_3)_4]$ (10)	5	K_3PO_4 (2)	DMF	1, 26	63	5	5
14 ^[c]	[Pd(PPh ₃) ₄] (10)	5	NaOAc (3)	DMF	1, 44	82	_	3
15 ^[c]	$[Pd(PPh_3)_4]$ (5)	2.5	NaOAc (3)	DMF	1, 24	77	6	3

[a] All reactions were carried out under argon using 1a (1.0 equiv), 2a (1.0 equiv), Pd catalyst, base, and solvent (c = 0.1 m) at 60°C for the indicated time (t_1); then, 3a (1.1 equiv) was added and the reaction mixture was heated at 110°C (t_2). [b] Only compound 5a formed, but its yield was not determined (n.d.). [c] 1.1 equiv 1a was used. dba=dibenzylideneacetone.

Other inorganic bases were examined, and NaOAc and KOAc were found to be superior to NaHCO₃ and K₃PO₄ (Table 1, entries 9-13), with three equivalents of NaOAc being optimal (entry 14). Among the palladium catalysts investigated (Table 1, entries 2–5), tetrakis(triphenylphosphine) was found to be the most effective (entry 2). DMF proved to be a better solvent than N,N-dimethylacetamide (DMA) or dimethyl sulfoxide (DMSO), while dioxane was inefficient in promoting the second step (Table 1, entries 6-8). The presence of copper iodide is essential to the Sonogashira coupling, but it retarded the carbopalladation/C-H functionalization process. The best Pd/Cu ratio was found to be 2:1 (Table 1, entry 2 vs. 9). Overall, under optimum conditions (5 mol % [Pd(PPh₃)₄], 2.5 mol% CuI, 3 equiv NaOAc, DMF, c = 0.1m; Table 1, entry 15), the reaction of 1a, 2a, and 3a provided the desired oxindole 4a in 77% yield. Under these conditions, the phenylpropiolamide 5a and the symmetrically substituted oxindole 4b were also isolated in yields of 6% and 3%, respectively.

To probe the scope and limitation of this new three-component reaction, a range of substituted aryl iodides and propiolamides were next examined (Scheme 2, Table 2). The outcome of this three-component reaction was found to be

Scheme 2. Anilides 1 used in Table 2.

only weakly influenced by the electronic properties of the aryl halides, and a variety of oxindoles substituted by an electrondonating or an electron-withdrawing group were readily synthesized in good to excellent yields (Table 2, entries 1–5). Whereas ortho-, meta-, and para-substituted aryl iodides can be used as the second aryl iodide (Table 2, entries 1, 6, and 8), only para- and meta-substituted aryl halides were well tolerated when used as the first aryl iodide added. The chlorine atom is compatible with this reaction sequence by providing a handle for further functionalization (Table 2, The N-(2-trimethylsilylethoxyand 8). methoxy)anilide 3b participated in this three-component reaction albeit the yield was slightly decreased (Table 2, entry 8). As an N-SEM group is readily removed, it constituted a route to N-unsubstituted oxindoles. Note that both

Scheme 3. Synthesis of symmetrically substituted 3-(diarylmethylene)oxindoles.

Table 2: Scope of the palladium-catalyzed three-component synthesis of 3-(diarylmethylene)oxindoles. [a] Structures of anilides 1 are given in Scheme 3.

Entry	1	Ar ¹ I	Ar ² I	Product	Yield [%] ^[b]	Entry	1	Ar ¹ I	Ar ² I	Product	Yield [%] ^[b]
1	la		NO ₂	MeO NO2 NeO Aa Me	77	9	1c	MeO	NO ₂	NO ₂ OMe OMe Aj	67
2	la	MeO	NO ₂	MeO NO2 OMe	82	10	la	NO ₂		MeO O ₂ N O ₂ N O ₂ N O ₃ N O ₄ N Me	< 10
3	1 a	O ₂ N	NO ₂	NO ₂ NO ₂ NO ₂ Ad Me	71	11	la	MeO		MeO OMe	69
4	1 a	CI	NO ₂	MeO NO ₂ CI	76	12	la		NO ₂	MeO NO2 N MeO MeO MeO MeO N MeO N MeO N MeO N MeO N Me	81
5	la	F_3C	NO ₂	MeO CF ₃ 4f Me	78	13	la		S I	MeO Neo An Me	61
6	la		F ₃ C	F ₅ C MeO NeO MeO MeO MeO MeO NeO MeO NeO NeO NeO MeO NeO NeO MeO MeO NeO NeO NeO NeO NeO NeO NeO NeO NeO N	66	14	la		H	MeO No	45
7	la	F ₃ C		MeO CF ₃	43	15	la	MeO		MeO No OMe	0
8	1Ь		CI	CI N 4i SEM	59	16	la		NO ₂	MeO NO2 NO4	51

[a] General conditions: 1a (1.1 equiv), Ar¹I (1.0 equiv), CuI (2.5 mol%), [Pd(PPh₃)₄] (5 mol%), NaOAc (3 equiv), DMF (c=0.1 m), 60°C; then Ar²I (1.1 equiv), 110°C. [b] Yield of product isolated after flash column chromatography.

the E- and the Z-oxindole can be prepared from the same starting materials by simply changing the order of addition of the two aryl halides (Table 2, entries 6 and 7).

To further exploit the generality of this catalytic MCR and because of the biological relevancy of heterocycles, we investigated the possibility of incorporating heteroaryl groups into the target compounds. As is seen from Table 2

(entries 11-16), 3-iodopyridine, 5-iodoindole, and 2-iodothiophene were appropriate starting materials for this MCR. Whereas 3-iodopyridine was a suitable reactant for both the Sonogashira coupling and the carbopalladation step, leading to compound 41 and 4m, respectively, 2-iodopyridine could only be used as a reaction partner in the Sonogashira coupling (Ar¹I). When it was used for the carbapalladation/direct C-H

Communications

functionalization sequence (Ar¹I), the expected oxindole **4p** was not formed. Most probably, the nitrogen atom of 2-pyridine is able to sequester the Pd^{II} species by forming a five-membered chelate after the carbopalladation step, thereby blocking the subsequent C–H activation process. Nevertheless, 2-thiophene was a suitable partner to furnish the oxindole **4n** in 61% yield (Table 2, entry 13). Finally, a symmetrically substituted oxindole, **4r**, was synthesized in 57% yield by simply conducting the reaction with two equivalents of 3-(trifluoromethyl)phenyl iodide (Scheme 3).

In summary, we have developed a flexible and stereocontrolled three-component synthesis of unsymmetrically substituted 3-(diarylmethylene)indolinones. The overall reaction involves a sequence of Sonogashira reaction/carbopalladation/C—H activation, and C—C bond formation and is catalyzed by a single Pd/Cu catalytic system. The ready accessibility of the starting materials and the generality of this process make it highly valuable in view of the synthetic and medicinal importance of these heterocycles.

Experimental Section

Typical procedure: Copper iodide (2.5 mol%), NaOAc (3 equiv), and $[Pd(Ph_3)_4]$ (5 mol%) were added to a degassed solution of anilide (1.1 equiv) and aryl iodide (1.0 equiv) in DMF (c=0.1m). The mixture was heated at 60°C for 1 h, and then the second aryl iodide (1.1 equiv) was introduced. After being stirred at 110°C for 28 h, the reaction mixture was quenched with saturated aqueous NaCl solution and extracted with EtOAc. The combined organic layers were washed with brine and dried (Na₂SO₄), and then concentrated. The residue was purified by flash chromatography to give the corresponding oxindole.

(3*E*)-5-methoxy-1-methyl-3-[(2-nitrophenyl)(phenyl)methylene]-1,3-dihydro-2*H*-indol-2-one (**4a**): 72 mg (77% yield); red solid; m.p. 197–200°C; 1 H NMR (300 MHz, CDCl₃): δ = 8.19 (dd, J = 8.2, 1.1 Hz, 1 H), 7.77 (dt, J = 7.5, 1.1 Hz, 1 H), 7.65 (dt, J = 8.2, 1.5 Hz, 1 H), 7.55 (dd, J = 7.5, 1.5 Hz, 1 H), 7.44–7.39 (m, 2 H), 7.36–7.31 (m, 3 H), 6.73 (dd, J = 8.5, 2.4 Hz, 1 H), 6.65 (d, J = 8.5 Hz, 1 H), 5.60 (d, J = 2.4 Hz, 1 H), 3.48 (s, 3 H), 3.17 ppm (s, 3 H); 13 C NMR (75 MHz, CDCl₃): δ = 166.0, 155.0, 149.4, 147.5, 137.6, 136.4, 136.2, 134.3, 131.3, 130.2, 129.9, 129.5, 127.6, 125.6, 125.0, 123.4, 114.1, 110.0, 108.17, 55.5, 25.9 ppm; HRMS: calcd for C_{23} H₁₈N₂O₄Na [M+Na] $^{+}$: 409.1164; found: 409.1182.

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3295